Docket No. 56778 (70820)



#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

APPLICANT:

Y. Sasaki et al.

U.S.S.N.:

10/019,743

GROUP:

1653

FILED:

December 28, 2001

EXAMINER: A. U. Desai

FOR:

PROCESS FOR PREPARING LH-RH DERIVATIVES

Mail Stop Amendment Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

#### **DECLARATION UNDER 37 CFR 1.131**

The undersigned declare as follows:

- 1. We are co-inventors of the above-identified patent application assigned to the Takeda Chemical Industries, Ltd.
- 2. Prior to December 18, 1998, we made and successfully isolated Purified leuprorelin or a salt thereof having a sum of all impurities of less than about 1% as disclosed and claimed in the above-identified patent application.
- 3. That the purified leuprorelin we made and isolated prior to December 18, 1998 also included purified compositions in which the content of 5-oxo-Pro-D-His-Trp-Ser-Tyr-D-Leu-Leu-Arg-Pro-NH-CH<sub>2</sub>-CH<sub>3</sub> or a salt thereof was about 0.3% or less.
- 4. That the purified leuprorelin we made and isolated prior to December 18, 1998 also included purified leoprorelin compositions in which the impurities were racemic isomers of the LH-RH derivatives and/or highly polar related substances.

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- 5. Attached as Exhibit 1 is a true and accurate copy of laboratory notebook records, as well as partial English translations of the notebook records, with dates deleted. The notebook records demonstrate that the purified leuprorelins as described in paragraphs 2, 3, and 4 above were made prior to December 18, 1998. Exhibit 1 shows, among other things, methods of characterization of the purified leuprorelin compositions, confirmation of reproducibility by scale-up, and analytical analysis for the purified leuprorelin as described in paragraphs 2, 3, and 4 above. On that Exhibit 1, the references to "S3-TAP and TAP-144" designate the particular purified leuprorelin compositions having less than 1 % impurities.
- 5. We hereby further declare that all statements made herein of our own knowledge are true and that all statements made on information and belief are believed to be true, and further that these statements are made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment or both (18 U.S.C. 1001), and that such willful false statements may jeopardize the validity of the above-identified application or any patent issued thereon.

Date: December 17, 2004

Yasuhiro Sasaki

Date: Decomber 17, 2004

Katsuii Shimizu



# Exhibit A

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## Partial English Translation of Laboratory Notebook Records

## Page 18 (12258-18)

Methane sulfonic acid x 1.25 Confirmation of inhibition of His effect and preparation at reaction temp. 15°C

1.	300	ml	4-neck	flask
----	-----	----	--------	-------

2. methane sulfonic acid 17.8+ 32.3 (1.25)

3. phenol 13.4 g

4. cooling S

5. NBS-TAP weighing

6. ditto charge (8°C or lower)

7. reaction S

(3 Hr)

8. finished

1. tare 513.39 g

2. including washing

3.

4. 8:42

5. 12.87 g (corr . to TAP 11.0 g)

6. 8:55 (7.0) - 9:07 (8.0)

7. 9:07 (9.0)

8.12:08

15. concentrated at 25°C of outer temp.

15. 15:10 (25)-15:40 (25)

615.16 613.96 f=0.9980

aqueous layer loss 0.0076g/H<sub>2</sub>O 0.07%

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## Page 19 (12258-19)

· MBS · TAP (dry)

plant product Lot076 (P: 92.87%)

potassium carbonate 0.25 min. additional dissolution (30.76 g)

· buffer pH  $8.21 \rightarrow 4.08$  (acetic acid 11.5 ml)

remainder MBS· TAP

0.0177%

0.0422 g/11.95 = 0.35%

reaction rate 4.23%

10.085/11.0 = 91.68%

reaction mixture

91.68%

before concentration

78.72%

after concentration

85.34%

after overnight

87.61%

/.58% 88.73%

related substances (USP method)

	D-Trp	D-Ser	D-His	L-Leu
reaction mixture	1.60	0.08	0.23	0.90
before conc.		<b>-</b>	0.25	0.76
after conc.	-	-	0.26	0.92
after overnight	-	-	0.24	0.93
refrigeration		0.01	0.25	0.94

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## Page 88 (10279-88)

Confirmation of reproducibility by scale-up (1)

For the purpose of reducing one column (HP-2MG→HP-20SS→LH-20) from the previous 4 columns (HP-20→ CH-23→HP-20→LH-20), the new process is confirmed by scale-up.

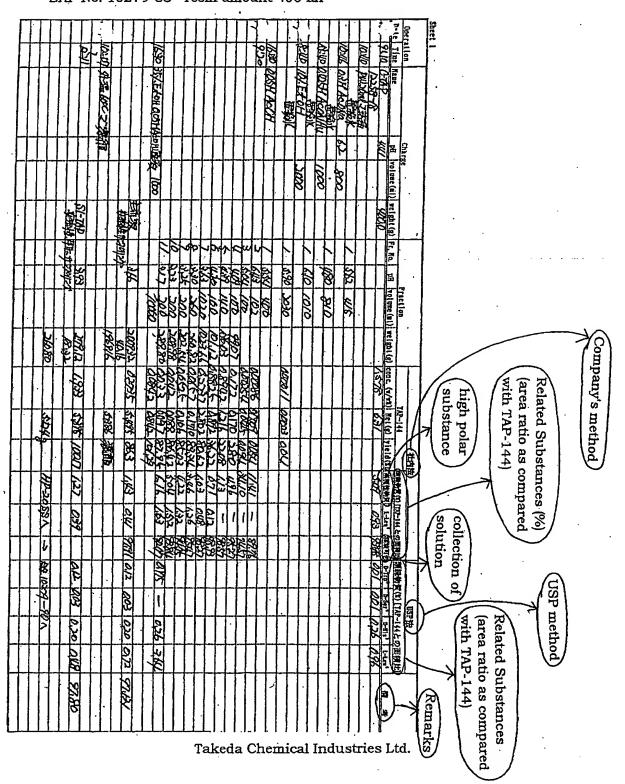
chromatographic purification by HP-2MG C-TAP→S1-TAP cf. detailed data sheet

			•
D-Trp3 form	D-Ser4 form	D-His2 form	L-Leu <sup>6</sup> form
0.01%	0.01%	0.26%	0.96%
0.12%	0.03%	0.20%	0.48%
area %	recove	ery	
93.98/77.43	88.739	%	
98.08/97.80	86.45	%	
	0.01% 0.12% area % 93.98/77.43	0.01% 0.01% 0.12% 0.03% area % recove .93.98/77.43 88.739	0.01% 0.01% 0.26% 0.12% 0.03% 0.20% area % recovery 93.98/77.43 88.73%

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Page 89 (10279-89)

AP-144 (500 Step) Column Chromatography Experimental Record HP-2MG 5°C passage (jacket) EXP No. 10279-88 resin amount 400 ml



## Page 90 (10279-90)

Purification of S1-TAP by HP-20SS S1-TAP 10279-88 HP-20SS passage temperature 15°C, resin amount 20% up 175 ml cf. detailed data sheet

## Results

S2-TAP	D-Trp <sup>3</sup> form	D-Ser <sup>4</sup> form	D-His² form	L-Leu form
	0.13	0.04	0.20	not detected
S2-TAP	area % 99.54/99.45	recovery 90.23		

Page 91 (10279-91)

AP-144 (500 Step) Column Chromatography Experimental Record HP-20SS Temp. 15°C

EXP No. 10279-90 resin amount 175 ml \*resin amount 20% up S1-TAP Lot 10279-88 Company's method with TAP-144) Related Substances (%) (area ratio as compared substance collection of USP method with TAP-144) Related Substances (area ratio as compared Remarks Takeda Chemical Industries Ltd.

## Page 96 (10279-96)

## Purification by LH-20

S2-TAP 10279-90

Passage conditions is the current method.

cf. detailed data sheet

## Results

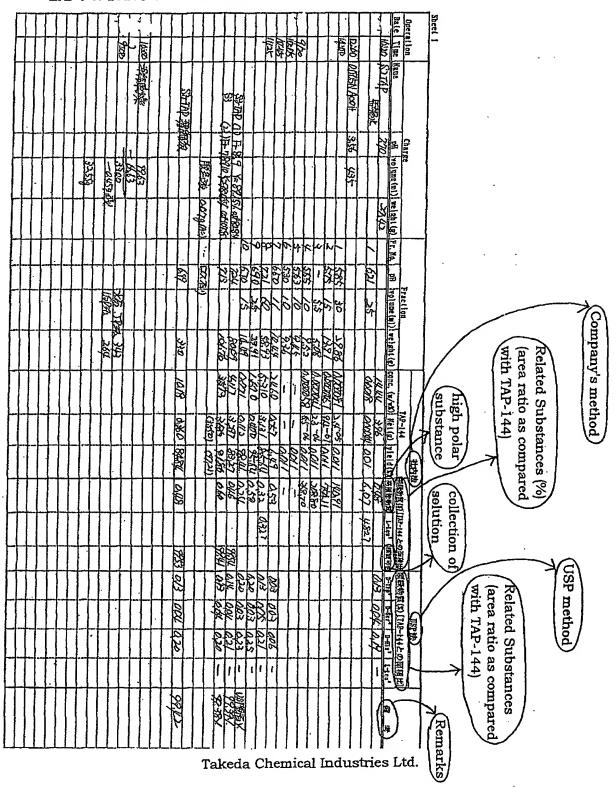
	D-Trp3 form	D-Ser4 form	D-His form	L-Leu <sup>6</sup> form
S3-TAP	0.13	0.04	0.20	$N \cdot D$
S4-TAP	0.13	0.04	0.20	$N \cdot D$
product	0.14	0.05	0.19	$N \cdot D$
	area %	recovery		
S3-TAP	99.54/99.46	90.23		
S4-TAP	99.54/99.42	97.21		
product	99.52/99.45			

Moisture Measurement

Page 97 (10279-97)

AP-144 (500 Step) Column Chromatography Experimental Record LH-20 room temp.

EXP No. 10279-96 resin amount 145 ml



## Summary of Records

Re: Details of obtaining a solution containing TAP-144 whose impurities content is 1% or lower and D-His<sup>2</sup> form content is 0.3% or lower

D-His<sup>2</sup> form content has been already reduced to 0.3% or lower by studying de-MBS reaction conditions before purification (C-TAP), and then, impurities in TAP-144 is reduced to 1.0% or lower by chromatographic purification.

Date	Steps	Experimental	Results	Notebook
L		method		No.
į	MBSTAP	Purpose: To confirm effect	USP method	12258-
	1	on control of D-His2 form.	area %	18 to 19
	C-TAB	Experimental method	D-His2 form:	
	(corr. to	Amount of MESE charge:	0.25%	
1	Example 1)	1.25-fold		
	Ì	Reaction time: 3 hrs		
		Reaction temp.: 11°C		<u> </u>
	C-TAP	Purpose: To purify C-TAP	USP method	10279-
. · .		by HP2MG.	area %	88 to 89
, ·	S1-TAP	Experimental method	S1-TAP	
	(corr. to	Passage temp.: 5°C (jacket)	D-His2 form:	
	Example 2)	Resin amount: 400 ml	0.26%→	
[ i		C-TAP amount: 6.31 g	0.20%	
		Solvent: 0.05M AcOH	TAP-144:	
			97.80%	
	S1-TAP	Purpose: To purify S1-TAP	USP method	10279-
	↓	by HP-20SS	area %	90 to 91
	S3-TAP	Experimental method	S3-TAP	
	(corr. to	Passage temp.: 15°C	D-His² form:	
	Example 3)	(jacket)	0.20%→	
		Resin amount: 175 ml	0.20%	
	•	C-TAP amount: 4.90g	TAP-144:	
		Solvent: 20% EtOH→	97.80%→	
		35% EtOH	99.46%	
	S3-TAP	Purpose: To purify S3-TAP	USP method	10279-
Í	↓	by LH-20	area %	96 to 97
	S4-TAP	Experimental method	S4-TAP	ļ
ŀ	(corr. to	(current method)	D-His² form:	ļ
	Example 4)	Passage temp.: room temp.	0.20%→	
- 1		Resin amount: 145 ml	0.20%	
		C-TAP amount: 3.96g	TAP-144:	İ
		Solvent: 0.005N AcOH	99.46%→	
			99.42%	

## Abbreviations in

Notebook	Examples in the	<b>Present Application</b>
----------	-----------------	----------------------------

MBSTAP 5-oxo-L-propyl-L-histidyl-L-tryptophyl-L-seryl-L-tyrosyl-

D-leucyl-L-leucyl-Nw-p-methoxybenzenesulfonyl-L-arginyl-

N-ethyl-L-prolinamide

C-TAP Aqueous solution (1) of Leuprolide S1-TAP Aqueous solution (2) of Leuprolide

S3-TAP Aqueous solution (3) of Leuprolide

S4-TAP Concentrated fractions containing leuprolide eluted with

a  $0.005 \, \text{M}$  aqueous solution of acetic acid